

## GYMNÁZIUM SOBRANCE

Projekt Erazmus Plus KA 101

### Inovatívnosť a flexibilita - záruka kyality vzdelávania

### Pracovný list

Predmet: chémia

Názov tematického celku:Spoznávame chemické reakcie v našom okolí

Názov učebnej látky: Chemické zlučovanie

Trieda: sekunda Dátum: 6. 8. 2019

### Worksheet

Lesson: Chemical synthesis of iron and sulfur

Few question for begining?

What is iron?
What is sulfur?
What's mean synthesis?
What is product of this reaction?

#### Iron

is a <u>chemical element</u> with symbol **Fe.** It is a <u>metal</u> and it is the fourth most common <u>element in the Earth's crust</u>. It is a heavy malleable ductile magnetic silver-white metallic element. Iron powder is grey. ......

### Sulfur

is a chemical element with symbol **S**. It is a non-metal and a bright yellow crystalline solid which is insoluble in water. Sulphur burns in air with a bright blue flame to produces sulfur dioxide which is toxic and use to desinfection against bacteria or fungi.

#### What we will need?

- Test tube
- Burner
- Chemical spoon
- Porcelain bowl
- Test tube holder
- Magnet
- Petri dish
- sulfur powder
- iron powder

### Instructions

- 1. 2 grams of sulfur powder mix with 3,5 grams of iron
- 2. Spatula the iron/sulfur mixture into a test tube
- 3. Heat it with flame of burner
- 4. The iron and Sulfur are now reacting together.





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- 5. The orange glow is indicated that this is an exotermic reaction (energy released to the surroundings
- 6. The black solid product is iron sulfide.

Whatch the experiment and fill up the table.

·	color	visual appearance	Magnetic properties	<u>yes/no</u>
sulfur				
iron				
mixture				
Mixture after heating				

<u>Chemical synthesis is a reaction in which two or more simple reactants join together into a one dificult product.</u>

(The production of chemical compounds by reaction from simpler materials)

 $R1 + R2 \longrightarrow P$ 

Iron + sulfur → Iron sulfide

## Questions

What is an iron?

What is a sulfur?

What's mean synthesis?

What is product of this reaction?

Copy the blue table page 27.